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#### Research article

# Characterization of hot-pressed biodegradable zinc-based nanocomposite implant materials reinforced with 10 wt% Mg, WE43, and AZ91



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# ABSTRACT

Compared to permanent orthopedic implants for load-bearing applications, biodegradable implants eliminate the necessity for surgical removal after the healing process. Furthermore, magnesium alloy powder reinforced zinc matrix implant materials have been produced to enhance the mechanical properties, biocompatibility, and a proper degradation rate with the growth rate of new bones. This study aims to fabricate Zn-10 wt% Mg, Zn-10 wt% WE43, Zn-10 wt% AZ91, and alloys along with pure Zn sample for control, using the powder metallurgy production method. In this context, hot pressing was applied to samples at 200 °C and 300 °C temperatures, under a constant pressure of 400 MPa to optimize the fabrication parameters. Scanning Electron Microscope (SEM), Energy Dispersive Spectrometry (EDS), Vickers macro- and micro-hardness test (HV), and X-Ray Diffraction Spectroscopy (XRD) analyses were performed to investigate the influence of press temperatures on the microstructure, elemental components, and mechanical properties of the fabricated samples. The microstructures of the zinc matrix nanocomposite samples reinforced with magnesium alloys predominantly consist of MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub>, and MgO phases dispersed within the refined zinc matrix. The obtained results indicate that Zn - Mg alloy nanocomposites hold significant potential as biodegradable orthopedic implant materials; however, it is possible to further improve the properties of the material by optimizing the production parameters. © 2025 The Authors. Published by Synsint Research Group.

## 1. Introduction

Biodegradable implants provide crucial advantages in load-bearing applications compared to conventional permanent implant materials. Biodegradable implants are engineered to gradually self-degrade once their supporting functions in the human body cease. This self-dissolving advantage negates the need for additional surgical intervention to remove the implant material, thereby reducing healthcare costs and minimizing the patient's post-operative recovery period [1–4]. As the global population ages, the demand for surgical reconstruction increases, creating a significant need for implant materials in surgical procedures. For instance, in the United States alone, approximately 100,000 anterior cruciate ligament (ACL) repair surgeries are performed annually [5]. Addressing this situation requires





## KEYWORDS

Zn-Mg alloy Nanocomposite Powder metallurgy Biodegradability Mechanical properties



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increased research and innovation projects aimed at the design and development of next generation biodegradable implant materials. Among various material's categories, biodegradable metallic materials are considered by researchers to be more suitable materials for implant applications, owing to their superior mechanical properties compared to polymeric alternatives.

Zinc is a crucial element, which has a vital function within the human body and is considered as one of the fundamental demanded components for human physiological processes [6]. This element is extensively distributed in body tissues, with approximately 85% found in muscle and bone, 11% is found in skin and liver, and the remaining portion is presented throughout the other tissues. In multicellular organisms, roughly 30-40% of the zinc element is stored in the nucleus, 50% is stored within the cytoplasm, organelles and special vesicles (such as those in digestive enzymes or hormone storage domain), while 10-20% are remnants in the cell membrane. Due to the important functions of zinc within the human body, the recommended daily consumption amount has been determined as 15 mg [7]. Additionally, zinc plays a critical role in the human body, being a vital constituent of many enzymes and proteins. The risk of zinc poisoning is extremely rare, considering its LD50 is 27 g/day. However, since the vomiting threshold is around 225-400 mg/day, ingestion of this amount is practically impossible. Zinc deficiency within the human body negatively affects growth, neuronal development, immune function, protein and DNA synthesis, wound healing, and many other vital biological processes [8, 9]. Considering the evaluations described, zinc is suggested to be a promising biodegradable material, leading to intensive research focused on developing the zinc-based biodegradable composite implant materials [10-16].

Zinc (Zn) and magnesium (Mg) have become as key focal point among various biodegradable metallic materials due to their critical roles in numerous biological functions and their biocompatibility characteristics [17, 18]. A major challenge associated with magnesium is its high susceptibility to corrosion, which occurs as a result of the rapid biological degradation process in the human body. The resulting rapid corrosion can lead to a deterioration of the implant material's mechanical integrity during the tissue healing process. Moreover, the corrosion rates of magnesium and its alloys can cause accelerated hydrogen evolution and alkalinization of body fluids, thus potentially adversely affecting homeostatic mechanisms and impeding the tissue healing process and leading to additional challenges of implant's biomedical applications [19-21]. Recent research has shown that zinc, being a more noble metal compared to magnesium, offers substantially higher corrosion resistance behavior without compromising both its mechanical integrity and biocompatibility characteristics [8, 12, 22].

The main limitations of pure zinc, as a biodegradable metallic material for implants production, are its low mechanical strength and plasticity. The tensile strength of cast zinc is less than 20 MPa and the strain value is 0.2% [8]. To improve the pure Zinc's mechanical properties, the alloying method is often preferred. The additive powder materials used in the alloying method play a crucial role in bone formation, mineralization and numerous enzymatic reactions occurring in the human body. Especially magnesium and its alloys (WE43 and AZ91) stand out as promising candidates for such biomedical applications. Many new magnesium alloys have been developed to solve the rapid corrosion problem of pure magnesium material. Regarding this context Mg, WE43, and AZ91 alloys have emerged as suitable candidates for implant material production due to their adequate mechanical

properties and lower corrosion rates compared to the most other magnesium alloys. The demanded biodegradable implant material's properties are improved by reinforcing the matrix material with WE43 and AZ91 powders, thus the additive alloys increase the potential of zinc-based biodegradable implant material to provide mechanical support during tissue healing processes. Furthermore, superior mechanical properties can be imparted to the samples by applying sintering heat treatment in addition to thermal deformation production method such as powder metallurgy [23–27].

This research article aims to design a zinc-based biodegradable implant material with enhanced mechanical characteristics by the powder metallurgy production method. Three groups of samples that contain 10 wt% magnesium alloy powder (Mg, WE43, and AZ91) reinforced zinc matrix biodegradable nanocomposite material were produced. To optimize the production parameters, the effect of temperature on the mechanical properties of the material was investigated by hot pressing the samples at 400 MPa pressure under the temperatures of 200 °C and 300 °C for determining ideal production conditions.

## 2. Materials and methods

The Zn powder used as the matrix of the implant nanocomposite material has a purity of 99% and a powder size of <45  $\mu$ m, supplied by TMC Powder Metal company. Reinforcement powders are pressed samples produced by powder metallurgy method using pure Mg powders with grain size of <63  $\mu$ m, AZ91 magnesium alloy powders with particle size of <150  $\mu$ m (composition by weight %, Al: 8.77; Zn: 0.74; Ni <0.01; Cu <0.01 and the remnant is Mg) supplied by Zhejiang Bainianyin Industry and Trade Company, and WE43 magnesium alloy powders with particle size of <45  $\mu$ m (composition by weight %, Y: 3.7–4.3; Zr: 0.4–1; Nd: 2–2.5; Gd: 0.1–1.9 and the remnant is Mg) supplied by Magnesium Elektron® Ltd.

Regarding the powders preparation step, Mg, WE43, and AZ91 powders were added to the samples at a 10 wt% ratio. The powder mixtures were initially homogenized in a turbula mixer for 1 hour. Sample production was carried out using a specially designed single-axis, 200 ton capacity hydraulic press device, which was used to press the samples at temperatures of 200 °C and 300 °C under the constant pressure of 400 MPa.

In order to evaluate the mechanical properties of the fabricated samples, density analysis was conducted using WSA-224T instrument

 
 Table 1. Codes and production parameters of Zn-based nanocomposite samples fabricated by powder metallurgy method.

	Pro	duction parar	neters
Sample code	Component (wt%)	Pressure (Mpa)	Temperature (°C)
A43	Pure Zn	400	300
BL42	10 Mg	400	200
BL43	10 Mg	400	300
DL42	10 WE43	400	200
DL43	10 WE43	400	300
CL42	10 AZ91	400	200
CL43	10 AZ91	400	300



Fig. 1. The SEM micrographs and EDS analyses results of the pure Zn reference sample, which coded as A43 and produced by powder metallurgy method.

in both air and water conditions. Furthermore, HV5 macro- and HV0.3 micro-hardness analyses were conducted using Emcotest Duravision 200 and Shimadzu HMV-2 instruments under the loads of 5 kg and 300 g, respectively. Through the characterization phase, density, macro-hardness and micro-hardness measurements, and the microstructural and compositional characterization of the nanocomposite materials were performed and analyzed.

#### 3. Results and discussion

#### 3.1. Microstructural and compositional characterization

The microstructural and elemental composition analyses of the pure Zn control sample, A43, that was produced by powder metallurgy method at a temperature of 300 °C under a constant pressure of 400 MPa, were examined using Scanning electron microscope (SEM) and Energy dispersive x-ray spectroscopy (EDS) and are presented in Fig. 1.

Moreover, the SEM micrographs and EDS analyses results for 10 wt% Mg reinforced Zn matrix nanocomposite demonstrated in Fig. 2. The EDS analysis revealed the significant changes in the phase elemental concentration of the samples reinforced with 10 wt% Mg, coded as BL42 and BL43, which were subjected to hot pressing at 200 °C and 300 °C variable temperature parameters, respectively, under a constant pressure factor of 400 MPa. As a result of hot pressing at 200 °C, intermetallics and phases such as MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub>, and MgO [28–31] nucleate and initiate to grow as relatively homogeneous and fine-grained precipitates within  $\alpha$ -Zn matrix, thus the intermetallics coarsen as a result of diffusion-driven growth due to the increase in the pressing temperature to 300 °C. The growth of intermetallics is encouraged due

to the increased diffusion as a result of the rise in the pressing temperature; however, it simultaneously induces phase separation, thermal distortion and the formation of micro-cracks specifically along grain boundaries, as evident in the SEM micrographs.

Fig. 3 indicates the SEM micrographs and EDS analysis results of the 10 wt% WE43 reinforced Zn matrix nanocomposite samples, coded as DL42 and DL43, which were hot pressed at 200 °C and 300 °C temperature and fabricated under 400 MPa constant pressure. The increase in pressing temperature from 200 °C to 300 °C significantly affects the distribution and recrystallization of MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub> intermetallics and MgO phase [28–32] in the  $\alpha$ -Zn matrix as shown in the SEM micrographs of the nanocomposite samples. The elevated temperature enhances the diffusion of magnesium into the zinc matrix, leading to the formation of coarser-grained MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub> intermetallic phases, alongside with more distinct phase boundaries between the phases. Additionally, the oxygen concentration increases from 13.34% in DL42 to 28.44% in DL43, indicating enhanced oxidation and MgO formation at 300 °C hot pressing temperature.

The EDS analysis of the 10 wt% AZ91 reinforced nanocomposite sample, coded as CL42, hot pressed at 200 °C under 400 MPa pressure, revealed moderate oxygen content (20.80%) promoted the formation of a homogeneously distributed MgO phase within the  $\alpha$ -Zn matrix. The presence of MgO indicates that limited diffusion occurred at the low pressing temperature. Upon reaching the pressing temperature of 300 °C during the production of the CL43 sample, it was observed that enhanced diffusion led to a higher oxygen concentration (28.93%) and a corresponding decrease in magnesium content, increasing in MgO phases. The SEM micrographs indicate the growth of MgZn<sub>2</sub> and

BL43 %10 Mg 30°C         Element B         Element Symbol         Element Conc. Conc. 6         C         Carbon Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 6         Magical Conc. Conc. 7         Magical Conc. 7         Magical Co	<b>BL42</b> %10 Mg 400 MPa 200°C	Element Number 6 8	Element Symbol C O	Element Name Carbon Oxygen	Atomic Conc. 6.846 14.987	Weight Conc. 2.400 7.000	USZEII WE USZEII WE USZEII WE
BL43 %10 Mg 400 Mpa 300°C         Element 6         C         Carbon         10.973         4.296           8         0         0xygen         15.133         7.892           12         Mg         Magnesium         52.048         41.259           30         Zn         Zinc         21.845         46.553	200 0	30	Zn	Zinc	29.289	55.900	200μm BI42 150μm BI42 30μm
BL43 %10 Mg 400 Mpa 300°C         Number         Symbol         Name         Conc.         Conc.           6         C         Carbon         10.973         4.296           8         O         Oxygen         15.133         7.892           12         Mg         Magnesium         52.048         41.259           30         Zn         Zinc         21.845         46.553		Element	Element	Element	Atomic	Weight	NgZall Ng2
%10 Mg         6         C         Carbon         10.973         4.296           400 MPa         8         O         Oxygen         15.133         7.892           12         Mg         Magnesium         52.048         41.259           30         Zn         Zinc         21.845         46.553	BL 43	Number	Symbol	Name	Conc.	Conc.	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	%10 Mg	6	C O	Carbon	10.973	4.296	en Nga
30 Zn Zinc 21.845 46.553	400 MPa 300°C	12	Mg	Magnesium	52.048	41.259	
		30	Zn	Zinc	21.845	46.553	

Fig. 2. The SEM micrographs and EDS analysis results of BL42 and BL43 samples with 10 wt% Mg content.



Fig. 3. The SEM micrographs and EDS analysis results of DL42 and DL43 samples reinforced with 10 wt% WE43 magnesium alloys powder.

 $Mg_2Zn_{11}$  intermetallics [28–31,33] along the  $\alpha$ -Zn matrix due to the elevated temperature. Additionally, macro- and micro-cracks were detected due to the increased intragranular stress concentration (Fig. 4). The general conclusion obtained from the SEM microstructures of the Zn-based nanocomposites indicates that the addition of pure Mg powder to the matrix results in various-shaped intermetallics. The WE43 reinforced nanocomposites demonstrate spherical intermetallics; however, the AZ91 reinforced nanocomposites exhibit a lamellar microstructure. It's important to note that the presence of carbon element in the EDS analyses of the samples is due to the bakelite material used in embedding the samples to ensure the stability of the samples during the metallography sample preparation and characterization stages, respectively.

## 3.2. Mechanical characterization

In this research paper, the standard Archimedes method was utilized to determine the density and the porosity of the reinforced Zn-based nanocomposites produced by the powder metallurgy method. Moreover, the density, porosity, and Vickers hardness values of fabricated samples are provided in Fig. 5.

The pure Zn sample, A43, which was hot pressed at 300 °C under

400 MPA pressure, exhibited the highest density of 92.16% with a low porosity of 7.84%, thus it was observed that the sample had a compact structure. In contrast to the A43 sample, the BL43 sample that reinforced with 10 wt% pure Mg powder and sintered after hot pressing procedure at 400 MPA and 300 °C indicated a minimum density value of 73.21% with a maximum porosity structure of 26.80%.

Vickers HV5 macro- and HV0.3 micro-hardness values explained the structural differences in the produced nanocomposite samples. While the HV0.3 micro-hardness values of the nanocomposite samples varied in the range of 52.9–72.4, the HV5 macro-hardness values were in the range of 53.5–73.0 [28, 34, 35]. Notably, the increased porosity of low density samples such as BL43 and DL42 caused them to exhibit lower hardness values. This characteristic behavior suggests that increased structural porosity reduces the integrity of the implant material, resulting in lower mechanical properties. The established correlation between porosity and hardness properties of hot-pressed samples by powder metallurgy method provides the basis for optimizing density, porosity and hardness of nanocomposite implant samples for their intended use by emphasizing the effect of microstructural properties on the mechanical performance of the fabricated material.

	Element	Element	Element	Atomic	Weight
	Number	Symbol	Name	Conc.	Conc.
CL42	8	0	Oxygen	20.786	11.700
400 MPa	12	Mg	Magnesium	63.231	54.100
200°C	13	Al	Aluminum	1.897	1.800
	30	Zn	Zinc	14.086	32.400
	L				
	-				
	Element	Element	Element	Atomic	Weight
	Number	Symbol	Name	Conc.	Conc.
CL43	6	C	Carbon	16.489	8.308
%10 AZ91	8	0	Oxygen	28.929	19.419
400 MPa	12	Mg	Magnesium	42.970	43.844
300°C		1		1	
	13	Al	Aluminum	2.123	2.402

Fig. 4. The SEM micrographs and EDS analysis results of CL42 and CL43 samples containing 10 wt% AZ91 magnesium alloys powder.



Fig. 5. The mechanical characteristic results of hot-pressed Zn-based nanocomposite samples.

## 3.3. Structural analysis

Fig. 6a–c shows the X-ray diffraction (XRD) analysis diagrams of the nanocomposite samples, which were conducted using a Bruker D8 Advance instrument. For the mentioned sample groups that hot-pressed

at different temperatures revealed that they are fully compatible with the standard PDF cards – ICDD 01-077-1177, ICDD 00-006-0664, ICDD 01-074-9943, ICDD 00-030-0794, ICDD 00-027-0759, and ICDD 03-065-1853 exhibit distinct diffraction peaks corresponding to the crystallographic planes of zinc.



Fig. 6. a-c) The X-ray diffraction (XRD) analysis diagrams of Zn-based nanocomposite samples.

The reinforcement of pure Mg and various Mg alloy powders into the zinc matrix caused the pure zinc peaks to shift to the left on the XRD diagrams. Furthermore, the obtained XRD analysis results confirmed that both pure zinc (Zn) and zinc oxide (ZnO) have a hexagonal crystal structure, which are the primary matrix phases. Significant reflections of Zn were determined in the reference pure zinc sample at the (111), (100), (101), and (102) planes corresponding to two theta (20) angle values of 27.361°, 39.802°, 44.014°, and 55.096° [36, 37]. Additionally, pronounced reflections were observed at the (002), (101), and (201) planes corresponding to two theta (20) angle values of 35.146°, 36.926°, and 70.464°, respectively. These reflections supported by narrow and intense diffraction peaks in all three sample groups confirm the presence of crystalline ZnO phase alongside the structural consistency and various phase stability of the hot pressed nanocomposite materials.

In addition to the primary ZnO phase, the formation of intermetallic compounds containing zinc (Zn) and magnesium (Mg), especially MgZn<sub>2</sub> and Mg<sub>2</sub>Zn<sub>11</sub>, was identified. The MgZn<sub>2</sub> phase, characterized by its hexagonal crystal structure was confirmed by the reflections of the (110), (200), (004), (210), and (211) planes corresponding to two theta (20) angle values of 34.311 °, 39.826 °, 42.164 °, 53.560 °, and 54.704 °, respectively. Similarly, the presence of Mg<sub>2</sub>Zn<sub>11</sub> phase exhibiting cubic crystal structure was shown by prominent diffraction peaks in (311), (222), (321), (400), (410), (500), and (510) planes corresponding to two theta (20) angle values of 34.743 °, 36.343 °, 39.312 °, 42.195 °, 43.694 °, 53.547 °, and 54.759 °, respectively. In addition to these intermetallic phases, the presence of magnesium oxide (MgO) phase with cubic crystal structure observed in SEM micrographs was confirmed by (222) plane reflection corresponding to two theta (20) angle values of 38.439 °, respectively. The evaluated XRD results are in full agreement with the findings reported in the literature [28, 30, 31], and the presence of MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub> and MgO phases was detected in addition to the Zn and ZnO primary matrix phase.

## 4. Conclusions

In this study, 10 wt% Mg, WE43 and AZ91 reinforced Zn based nanocomposites were successfully synthesized by powder metallurgy method, then the effects of 200 °C and 300 °C pressing temperatures under 400 MPa constant pressure on the microstructural and mechanical characteristics of the samples were investigated. The results obtained showed that increasing the pressing temperature and incorporating reinforcement powders enhanced density, hardness, and intermetallic phase formation. Samples processed at 300 °C exhibited superior mechanical integrity, with the formation of MgZn<sub>2</sub>, Mg<sub>2</sub>Zn<sub>11</sub> and MgO phases affected the microstructural characteristics. The Vickers macro- and micro-hardness values range between 52.9 HV and 73.0 HV, depending on porosity and intermetallic distribution. Therefore, the nanocomposites hot pressed under 400 MPa pressure and 300 °C temperature demonstrated the most promising mechanical properties for implant applications, highlighting their potential suitability for biomedical use. Future research should focus on investigating biodegradation and corrosion resistance through in-vitro and in-vivo studies, as well as further optimizing processing parameters to enhance performance.

## **CRediT** authorship contribution statement

**Onur Fevzi Kevenlik:** Investigation, Conceptualization, Methodology, Data curation, Project administration, Writing – original draft.

Shanli Salahi: Investigation, Conceptualization, Methodology, Data curation, Writing – original draft.

Yiğit Yalçın: Investigation, Methodology, Data curation.

Hanifi Çinici: Conceptualization, Data curation, Validation, Writing – review & editing.

**Recep Çalın:** Project administration, Supervision, Funding, Writing – review & editing.

#### Data availability

The data underlying this article will be shared on reasonable request to the corresponding author.

## **Declaration of competing interest**

The authors declare no competing interests.

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